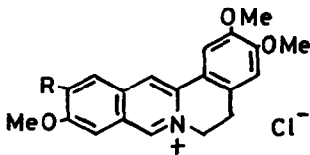


A NOVEL AMINATION OF ISOQUINOLINIUM SALTS VIA NUCLEOPHILIC
SUBSTITUTION REACTION

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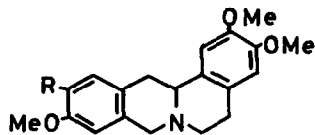
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The Smiles rearrangement at the C₉-position of protoberberinium salt¹ prompted us to examine a substitution reaction at that position. We now wish to report a new and unusual amination of 6- or 8-alkoxyisoquinolinium salts. Treatment of such salts with primary amines produces 6- or 8-aminoisoquinolinium derivatives. The reaction proceeded in refluxing ethanol or in refluxing methanol for 2 to 72 hours using 4 to 40 folds moles of amine, e.g., a mixture of dehydrocorydaline chloride(1a)(0.2 g), ethanolamine(2 ml) and ethanol(6 ml) was refluxed for 2 hr to give 9-(2-hydroxyethyl)amino derivative(2a)², mp 242-245°(dec), in 90% yield, whose bromide(2b), mp 233-236°(dec), was identified with the Smiles rearrangement product(2b)¹. Other examples of this reaction are shown in TABLE. Using benzylamine, 2,3,10,11-tetramethoxy-5,6-dihydrodibenzo[a,g]quinolizinium chloride(3), mp 220-222°(dec), obtained from dl-xylopinine(5a)³ gave 11-benzylamino derivative(4), mp 245-250°(dec). Reduction of 4 with sodium borohydride in methanol followed by catalytic debenylation led to the known 11-amino-tetrahydroprotoberberine derivative(5b)⁴, mp 225-228°.



3 R=OMe

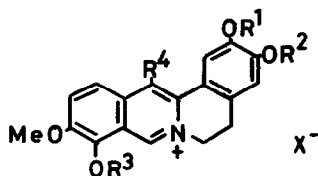
4 R=NHCH₂C₆H₅



5a R=OMe

5b R=NH₂

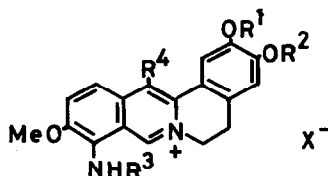
The C₉- and C₁₁-positions of protoberberinium salts are respectively corresponding to the C₆- and C₆-positions of isoquinolinium salts(6) being as an essential partial structure for this amination. These positions are activated by the presence of C=N⁺ group and the C₆-position of 6 is considered to be in less steric hindrance than the C₈-position of 6.



1

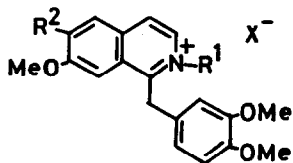
	R ¹	R ²	R ³	R ⁴	X
<u>1a</u>	Me	Me	Me	Me	Cl
<u>1b</u>	Me	Me	Me	H	CH ₃ SO ₄
<u>1c</u>	-CH ₂ -	Me	Me	Me	Cl
<u>1d</u>	-CH ₂ -	Me	H	H	Cl
<u>1e</u>	Me	Me	Pr	Me	Cl
<u>1f</u>	Me	Me	By	Me	Br
<u>1g</u>	Me	Me	H	Me	Cl

Me: CH₃
 Pr: CH₂CH₂CH₃
 By: CH₂C₆H₅




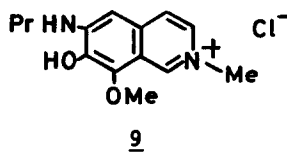
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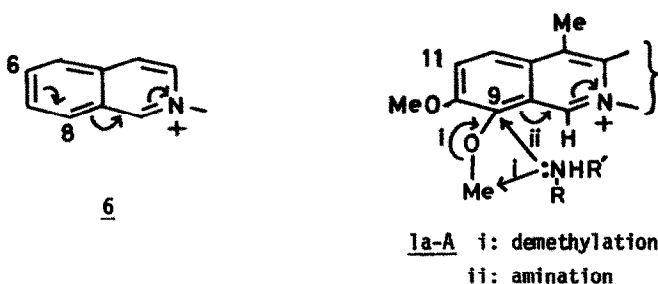
	R ¹	R ²	R ³	R ⁴	X
<u>2a</u>	Me	Me	CH ₂ CH ₂ OH	Me	Cl
<u>2b</u>	Me	Me	CH ₂ CH ₂ OH	Me	Br
<u>2c</u>	Me	Me	CH(CH ₃)CH ₂ CH ₃	Me	Cl
<u>2d</u>	Me	Me	CH ₂ CH ₂ NH ₂ ·HCl	Me	Cl
<u>2e</u>	Me	Me	Pr	Me	Cl
<u>2f</u>	Me	Me	By	Me	Cl
<u>2g</u>	Me	Me	CH ₂ By	Me	Cl
<u>2h</u>	Me	Me	By	H	CH ₃ SO ₄
<u>2i</u>	-CH ₂ -	CH ₂ CH ₂ OH	Me	Me	Cl
<u>2j</u>	-CH ₂ -	By	H	H	Cl



7

	R ¹	R ²	X
<u>7a</u>	Me	OMe	I
<u>7b</u>	By	OMe	Br
<u>7c</u>	Me	NHPr	I
<u>7d</u>	Me	NHCH ₂ CH ₂ OH	I
<u>7e</u>	Me	NHBy	Br
<u>7f</u>	By	NHBy	I
<u>7g</u>	Me		I



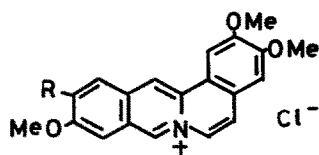


To test the generality of this amination, N-methylpapaverinium iodide(7a) was treated with propylamine. 6-Propylamino derivative(7c), mp 210-215°(dec), was obtained in 90% yield. Other amino-papaverine derivatives obtained by this route are indicated in TABLE. Amination of 0-methyltarconine chloride(8)⁵ with propylamine afforded predominantly one product(9), characterized as picrate, mp 166-168°, $\nu_{\text{NH}}^{\text{CHCl}_3}$ 3320 cm^{-1} , $\nu_{\text{OH}}^{\text{CHCl}_3}$ 3400 cm^{-1} , in 85% yield, in which the methylenedioxy group of 8 was cleaved to form an ortho-hydroxypropylamino group.

On the other hand, an attempted amination of 1a with pyrrolidine gave 9-desmethyldehydro-corydaline(1g)¹ in 90% yield besides N-methyl-pyrrolidine whose picrate was identified with an authentic sample. Other secondary amines and tert.-butylamine reacted with 1a as well as pyrrolidine. Using sec.-butylamine, 1a afforded two products, 9-sec.-butylamino derivative(2c), mp 238-240°(dec)(yield, 20%) and 1g(64%). From these results, it depends on the steric factor of the using amine whether the nucleophiles attack the C₉-position or the CH₃(Me)-group of C₉-OMe group of 1a as shown in 1a-A. Amination of 7a with pyrrolidine gave 6-pyrrolidino derivative(7g), mp 235-238°(dec), in 45% yield because of the less steric hindrance at that position as described above.

In a related experiment, norcoralyne chloride(10)³ was refluxed in methanol with benzylamine for 72 hours to generate predominantly one product(11), mp 245-247°(dec), in 85% yield. Reduction of 11 as in the case of 4 afforded 5b confirming the aminated position.

Further investigation on this amination is in progress.



10 R=OMe

11 R=NHCH₂C₆H₅

TABLE 2

Starting Salt	Amine	Solvent/ Reaction Time	Product	Yield	mp °C(dec)
<u>1a</u>	H ₂ NCH ₂ CH ₂ NH ₂	EtOH/ 2 hr	<u>2d</u>	90%	187-190
<u>1a</u>	PrNH ₂	MeOH/ 3 hr	<u>2e</u>	90%	214-215
<u>1a</u>	ByNH ₂	EtOH/ 5 hr	<u>2f</u>	86%	227-228
<u>1a</u>	ByCH ₂ NH ₂	EtOH/ 4 hr	<u>2g</u>	87%	205-206
<u>1b</u>	ByNH ₂	EtOH/ 2 hr	<u>2h</u>	85%	186-187
<u>1c</u>	H ₂ NCH ₂ CH ₂ OH	EtOH/50 hr	<u>2i</u>	60%	250-255
<u>1d</u>	ByNH ₂	MeOH/72 hr	<u>2j</u>	60%	269-271
<u>1e</u>	H ₂ NCH ₂ CH ₂ OH	EtOH/ 3 hr	<u>2b</u>	67%	233-236
<u>1f</u>	H ₂ NCH ₂ CH ₂ OH	EtOH/ 4 hr	<u>2b</u>	67%	233-236
<u>7a</u>	H ₂ NCH ₂ CH ₂ OH	EtOH/ 7 hr	<u>7d</u>	90%	215-217
<u>7a</u>	ByNH ₂	EtOH/ 6 hr	<u>7e</u>	90%	225-230
<u>7b</u>	ByNH ₂	EtOH/ 8 hr	<u>7f</u>	70%	140-145

References

- 1) S. Naruto, H. Mizuta, J. Nakano and H. Nishimura, Tetrahedron Letters, preceding paper in this issue.
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